

The Calvin Cycle

Introduction

- When **sunflower plants** are **illuminated** in the presence of **labelled CO₂**, it is found that of **7.87mg** of absorbed **CO₂**, **68%** ends up in **sucrose** and **23%** ends up in **starch**.
- The **dark reactions** are the second part of **photosynthesis** which **use** the **ATP** and **NADPH** produced by the light-reactions to reduce carbon atoms from their **fully oxidised state** as **CO₂** to a more **reduced state** such as **hexose**.
- In contrast to **gluconeogenesis**, where the energy ultimately comes from the catabolism of other fuels, the Calvin cycle can use energy from sunlight.
- The Calvin cycle takes place in the **stroma** of **chloroplasts**.

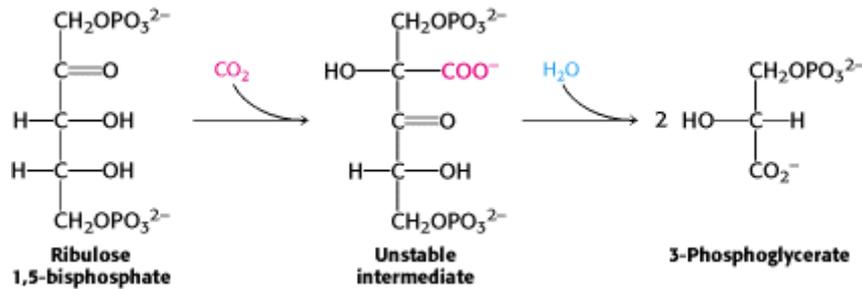
The Calvin Cycle

The Calvin Cycle consists of **three stages**:

- 1) The **fixation** of **CO₂** by **ribulose 1,5-bisphosphate** to form **two** molecules of **3-phosphoglycerate**.
- 2) The **reduction** of **3-phosphoglycerate** to form **hexose sugars**.
- 3) The **regeneration** of **ribulose 1,5-bisphosphate** so that **more CO₂** can be fixed.

Stage (1)

CO₂ condenses with **ribulose 1,5-bisphosphate** and forms an **unstable intermediate**, which is **rapidly hydrolysed** to **two molecules** of **3-phosphoglycerate**:



This **highly exergonic reaction** is catalysed by **ribulose 1,5-bisphosphate carboxylase/oxygenase** (usually called **rubisco**), an enzyme located on the **stromal side** of the **thylakoid membrane**. This reaction is the **rate-limiting step** in **hexose synthesis**.

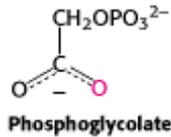
Rubisco is the **most abundant** enzyme, and probably protein, in the **biosphere**, probably because it's **so slow**. It is a **hexadecamer** of **eight large** (containing the active site) and **eight small** subunits.

Initial measurements of K_m were **so large** that the atmospheric concentration of CO_2 would have had to be incredibly high for the enzyme to operate at half its maximal velocity. We now know that the enzyme has to be **activated** by Mg^{2+} ions and an **alkaline pH**, and has to be **reduced**. The steps in the **preparation** of the enzyme are:

- 1) A CO_2 binds to the **uncharged ϵ -amino group** of **lysine 201** to form a **carbamate**. This is catalysed by **rubisco activase** [though this also occurs *slowly* when uncatalysed].
- 2) The Mg^{2+} can then bind to this **negatively charged carbamate**.
- 3) The Mg^{2+} is also bound to the enzyme via a **glutamate residue** and an **aspartate residue**.

The **ribulose 1,5-bisphosphate** then binds to the Mg^{2+} , where it is **readily deprotonated** [hence the need for alkaline conditions] to form an **enediolate intermediate**. This readily couples to CO_2 .

However, note that the enzyme has **no binding site for the CO_2 molecule to be added to ribulose 1,5-bisphosphate**. Therefore, it is unable to discriminate between CO_2 and O_2 , and the **enediolate intermediate** is sometimes attacked by O_2 . In such a case, as well as **3-phosphoglycerate**, a molecule of **phosphoglycolate** is produced:



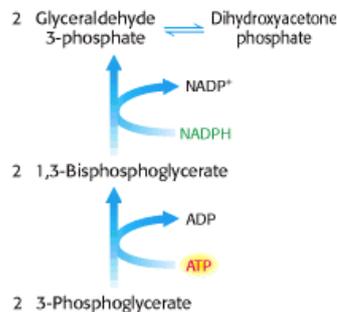
This is problematic, because **phosphoglycolate** is *not* a very versatile metabolite, and this it results in the **loss of CO₂** from the cycle. The cell is partially successful in recovering the carbon by turning **two** such molecules into **one** molecule of 3-phosphoglycerate – a 75% recovery.

Note, however, that since **rubisco** requires **binding of CO₂** to be bound, it will *not* simply mass-produce **phosphoglycolate** if **CO₂** is lacking. The process will simply **pause**.

This process is called **photorespiration**, because **O₂** is consumed and **CO₂** is released, without the production of any **energy-rich** metabolite. This imperfection in rubisco exists because it **evolved** before **oxygen** became **abundant** in the atmosphere. Evolutionary processes, however, *have* enhanced rubisco to a certain extent – the rubisco of **higher plants** is **eightfold** as specific for **carboxylation** than that of **photosynthetic bacteria**.

Stage (2)

Two reactions then convert these to **glyceraldehyde 3-phosphate**:



Which can then be fed into the **gluconeogenic pathway**. The two enzymes are **similar** to those in **glyconeogenesis**, apart from the fact that they use **NADPH** instead of **NADH**.

Stage (3)

Finally, **ribulose 1,5-bisphosphate** must be **regenerated**:

- 1) A series of **transformation reactions** catalysed by **transketolase** and **aldolase** generate **ribulose 5-phosphate**.
- 2) This is then **phosphorylated** by **phosphoribulose kinase** and **ATP** to re-form **ribulose bisphosphate**.

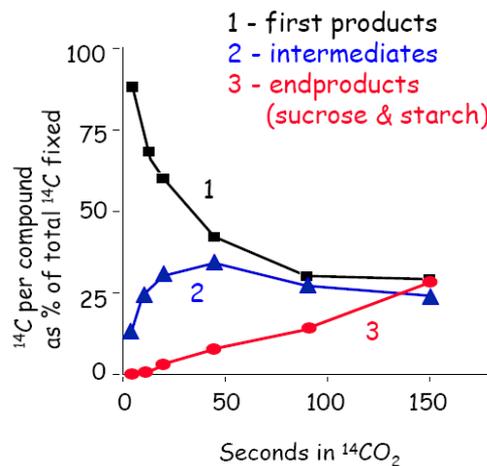
The **stoichiometry** of these reactions is:

- **3 CO₂** are **fixed**, which results in **SIX 3-carbon sugars**. This uses **6 ATP** and **6 NADPH**.
- The **condensation of sugar phosphates** in **different combinations** results in **THREE 5 carbon sugars** and **ONE 3 carbon sugar**.
- The **5 carbon sugars** are **phosphorylated** with **ATP** to **regenerate RuBP**.
- The **3 carbon sugar** (a **triose phosphate**) is **exported** to the **cytosol** to make **sucrose**.
- Some carbon remains in the **chloroplast** to make **starch**.

In total, therefore, **THREE ATP** and **TWO NADPH** are needed to bring each **CO₂** to **hexose level**.

Experimental Evidence

Experiments were carried out with **radiolabelled carbon compounds**, to establish how **sugar monomers** are made from **CO₂**. The following results were obtained:



Where:

- was found to be **3-phosphoglycerate**.
- was found to consist of various **intermediates** *formed* from **1** and **generating 3** (C3, C4, C5, C6 and C7 phosphorylated sugars). The *rates* at which these were labelled was so **similar** that the **sequence** of labelling could not be determined directly.
- were **sucrose** and starch.

Further progress came from the **determination** of the **distribution of ¹⁴C within the individual molecules**:

- Within **3-phosphoglycerate (3-phosphoglyceric acid)**

	% of label after	
	5 seconds	30 seconds
$\begin{array}{c} \text{CH}_2\text{OPO}_3^{2-} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{CO}_2^- \end{array}$	2.5	25
$\begin{array}{c} \text{HO}-\text{C}-\text{H} \\ \\ \text{CO}_2^- \end{array}$	2.5	25
CO_2^-	95	50

This established that

- a. The **CO₂** is **incorporated** into the **carboxyl group** of **phosphoglycerate**.
 - b. The fact that a label rapidly appeared in the other carbons established the existence of a **cyclic pathway** whereby the **acceptor** of the **CO₂** was formed from the **product** of the **carboxylation [3-phosphoglycerate** itself].
- Within one of the **C6 sugar phosphates, fructose 6-phosphate**

	% of label after 5 seconds
$\begin{array}{c} \text{O} \quad \text{CH}_2\text{OH} \\ \diagdown \quad / \\ \text{C} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OPO}_3^{2-} \end{array}$	3
$\begin{array}{c} \text{O} \\ \\ \text{C} \\ \\ \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OPO}_3^{2-} \end{array}$	3
$\begin{array}{c} \text{HO}-\text{C}-\text{H} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OPO}_3^{2-} \end{array}$	43
$\begin{array}{c} \text{H}-\text{C}-\text{OH} \\ \\ \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OPO}_3^{2-} \end{array}$	42
$\begin{array}{c} \text{H}-\text{C}-\text{OH} \\ \\ \text{CH}_2\text{OPO}_3^{2-} \end{array}$	3
$\text{CH}_2\text{OPO}_3^{2-}$	3

This **strongly suggested** that the two 3-carbon compounds formed from **3-phosphoglycerate** were **combined** to give a **6-carbon sugar phosphate**.

Sucrose & Starch

Sucrose and **starch** are the major **carbohydrate stores** in plants – the former is useful for **storage** (we use glycogen), the latter for **transport** (we use glucose):

- **Starch** is **synthesised** and **stored** in the **chloroplast**. It is a **polymer** of **glucose residues**, but it is less branched than **glycogen** (its animal counterpart) because it contains a smaller proportion of **α -1,6-glycosidic linkages**. Another difference is that **ADP-glucose** rather than **UDP-glucose** is its precursor.
- **Sucrose**, however, is a **disaccharide** synthesised in the **cytoplasm**. Since plants are unable to transport **hexose phosphates** across the cytoplasm, they transport instead **triose phosphates** such as **glyceraldehyde 3-phosphate** into the cytoplasm, in exchange for phosphate through abundant **phosphate translocators**. These then form **fructose 6-phosphate** which joins the **glucose** unit of **UDP-glucose** to form **sucrose 6-phosphate**. The hydrolysis yields **sucrose**, which is a **readily transportable** and **mobilizable** sugar.

Dependence on Environmental Conditions

In general, **photosynthesis** takes place during the day, whereas **catabolic metabolism** takes place at night. How are they **co-ordinately controlled**?

- We mentioned before that the **rate limiting step** in the dark reactions was the action of **rubisco**. Now, it turns out that in the presence of light:
 - The **pH increases from 7 to 8** in the **stroma** (due to **proton pumping**).
 - Levels of **Mg²⁺ rise** (to “balance” the **proton pumping**).

Both of these **favour** the formation of the **carbamate** necessary for enzyme activity. In the **dark**, this is **unlikely to happen**.

- **Thioredoxin** plays a **central role** in the regulation of the **calvin cycle**. The **presence of NADPH** and **reduced ferredoxin** are **good signals** that

conditions are **right** for **biosynthesis**. One way this is conveyed to **biosynthetic enzymes** is through **thioredoxin**, a 12-kd protein that contains **neighbouring cysteine residues** that cycle between a **reduced sulfhydryl form** and an **oxidised disulfide form**. The **oxidised form** can **active enzymes** by **reducing disulfide bridges** that **control their activity**, and **inhibit** several **degradative enzymes** in the same way.

In the **chloroplast**, **oxidised thioredoxin** is **reduced by ferredoxin** by an enzyme called **ferredoxin-thioredoxin reductase**. The enzyme contains a 4Fe-4S cluster that couples *two* one-electron oxidations of reduced ferredoxin to *one* two-electron reduction of thioredoxin.

- In the dark **phosphoribulose kinase** and **glyceraldehydes 3-phosphate dehydrogenase** are **inhibited** by **association** with a **small protein** called **CP12**. **NADPH** *disrupts* this **association**, leading to the release of the **active enzymes**.
- The **C₄ pathway** in **tropical plants** accelerates photosynthesis by **concentrating CO₂**. Four carbon compounds such as **oxaloacetate** and **malate** carry CO₂ from **mesophyll cells**, which are in contact with air, to **bundle-sheath cells**, which are the **major site of photosynthesis**. ATP is used in the process. This is important at **high temperature**, when the **oxygenase** action of **rubisco** becomes more and more significant.

The **pathway** can also be used to **temporally** rather than **spatially** separate the **absorption** of **CO₂** and its **utilization** in plants growing in **arid environments**, whose **stomata** have to **close** during the day and therefore cannot absorb CO₂. In such plants, C₄ sugars are made from **absorbed CO₂** at **night**, and this **CO₂** is **released** during the day.